

This thesis deals with the experimental and theoretical investigations to understand the atomic ordering and site preference of neighboring elements in Cu-rich intermetallic phases. The intermetallic compounds were prepared by the traditional high-temperature solid-state synthesis starting from highly pure constituent elements. Multiple structural characterization methods such as X-ray diffraction, neutron diffraction and energy dispersive X-ray spectroscopy analysis were employed in the experimental portion of this work. To complement the information from synthetic experiments, density functional theory (DFT) based first-principles computational codes such as tight-binding linear muffin-tin orbital employing atomic sphere approximation (TB-LMTO-ASA) and Vienna ab-initio simulation package (VASP), were used for electronic structure calculations.

Unusual atomic ordering and site preference of two neighboring transition metals Cu and Zn in  $\text{Cu}_6\text{Zn}_2\text{Sb}_2$  [ $P4/nmm$ ; 129] were investigated by combined experimental single crystal X-ray and neutron powder diffraction as well as the first-principles electronic structure calculations. The atomic ordering of Cu and Zn, previously observed in  $\beta'$ -CuZn, was found to persist upon the addition of Sb. The structure is built up by the sequential arrangement of  $\frac{2}{\infty}$   $[\text{Cu}_3\text{Sb}]$  slabs and sheets of Zn atoms or as stacking of  $\text{Cu}_2\text{Sb}$ -unit cells interleaved with  $\beta'$ -brass layers along  $c$ -direction. The specific substitution pattern of Zn to Cu site leads to the observed phase width from  $\text{Cu}_5\text{Zn}_3\text{Sb}_2$  to  $\text{Cu}_6\text{Zn}_2\text{Sb}_2$ .

Ternary Cu-Zn-X (X = Sn, In) compounds adopt the  $\gamma$ -brass structure type and exist in an extended homogeneity range, and obey the Hume-Rothery valence electron concentration rule. X-ray diffraction analyses confirm the site-specific Sn and In substitution in  $\gamma$ -brass structures.  $\gamma$ -Brass type phases with  $P$ -cell are formed in the Cu-Zn-Sn ternary intermetallics; whereas both  $I$ - and  $P$ -cell cubic  $\gamma$ -brass type phases exist for the Cu-Zn-In system.

$\gamma$ -Brass type phase  $\text{Cu}_{41}\text{Sn}_{11}$  [ $F\bar{4}3m$ ; 216] in the binary Cu-Sn system has been reinvestigated. These structures are constructed by four independent 26-atom  $\gamma$ -clusters and centered at the four high symmetry points of the unit cell. Electronic structure calculation shows that the phase is stabilized by the Hume-Rothery mechanism.

Ternary intermetallic compound  $\text{Cd}_2\text{Cu}_3\text{In}$  [ $P4_132$ ; 213] that is an ordered substitution variant of cubic  $\text{MgCu}_2$  structure type has been uncovered. The unexpected atomic ordering of two neighboring elements Cd and In has been investigated by first-principles DFT calculations, which identify the In-Cu and Cd-Cu bonding effect on the unusual observed ordering of In and Cd in the structure of  $\text{Cd}_2\text{Cu}_3\text{In}$ . The structure is described as a three-dimensional network of vertex sharing  $[\text{Cu}_3\text{In}]$  tetrahedra, whereas Cd atoms form a diamond-like network.

**Keywords:** Intermetallics; Gamma brass; Atomic ordering; Site preference; X-ray diffraction; Electronic structure calculations.