## Abstract

We have optimized various process parameters to synthesize CuO, LaFe<sub>0.8</sub>Co<sub>0.2</sub>O<sub>3</sub> (LFCO), ZnO (Z), LFCO – ZnO (LZ) composite, LFCO/ZnO (L/Z) and CuO/ZnO (C/Z) bilayer p-n heterojunction thin films using chemical solution deposition route. We have studied the ethanol and acetone sensing characteristics of CuO thin film and applied the gas diffusion theory to ptype CuO thin films of thickness 120-288 nm. We found that the gas response first decreased with increasing thickness and subsequently exhibited a volcano-shaped behaviour with further increase in film thickness for CuO thin films. We have developed the LFCO thin-film sensor for selective CO sensing applications assuming the efficient catalytic activity of the LFCO perovskite surface and the skillful microstructural control of thin-film morphology. We found that for film thickness in the range 92 - 432 nm, the 243 nm thin film exhibited the maximum response % to CO at 225°C and was also selective to CO in the studied temperature range. We have compared the CO and CO<sub>2</sub> sensing behaviour in LZ composite and L/Z p-n heterojunction bilayer thin films. We found that the LZ composite thin film is sensitive to both CO and CO<sub>2</sub> over a wide temperature range of 150-350°C whereas the L/Z sensor is selective to CO<sub>2</sub> between 150 and 300°C. The cross-sensitivity of CO and CO<sub>2</sub> in the LZ sensor was addressed via principal component analysis (PCA) of the resistance transients and simulation of conductance transients. The selectivity to CO<sub>2</sub> in the L/Z sensor is attributed to the formation of a p-n heterojunction between LFCO and ZnO. We have also studied the adsorption characteristics of CO<sub>2</sub> and C<sub>4</sub>H<sub>10</sub> gases on the C/Z sensor surface. We found that CO<sub>2</sub> follows the extended Freundlich model whereas C<sub>4</sub>H<sub>10</sub> obeys the extended Langmuir model of adsorption. From theory, we have calculated the activation energy, E<sub>A</sub> and heat of adsorption, Q for the test gases. These values for CO<sub>2</sub> and C<sub>4</sub>H<sub>10</sub> are found to vary considerably in the range of concentrations studied. We have also fabricated a portable and economical gas-sensing prototype. The efficiency of the prototype was tested in the laboratory as well as in the field.

**Keywords:** Thin film gas sensor, gas diffusion theory, CO sensing, CO<sub>2</sub> sensing, adsorption theory, portable sensing prototype